

Large Kinetic Isotope Effects for the Protonolysis of Metal-Methyl Complexes Are Not Reliable Mechanistic Indicators

Valerie J. Scott, Jay A. Labinger* and John E. Bercaw*

Supporting Information

Contents

1. Table of measured KIEs for the protonolysis of M-Me complexes at different temperatures using different methods.	2
2. General considerations for Arrhenius plots.....	3
3. Arrhenius plot and equation for (cod)PtMe ₂	4
4. Arrhenius plot and equation for (dppe)PdMe ₂ with TFA.....	5
5. Arrhenius plot and equation for ^t Bu ₂ Cp ₂ ZrMe ₂	6
6. Arrhenius plot and equation for ZnMe ₂	7
7. Table of parameters and R ² values for Arrhenius plots for complexes.....	8

1. Table of measured KIEs for the protonolysis of M-Me complexes at different temperatures using different methods.

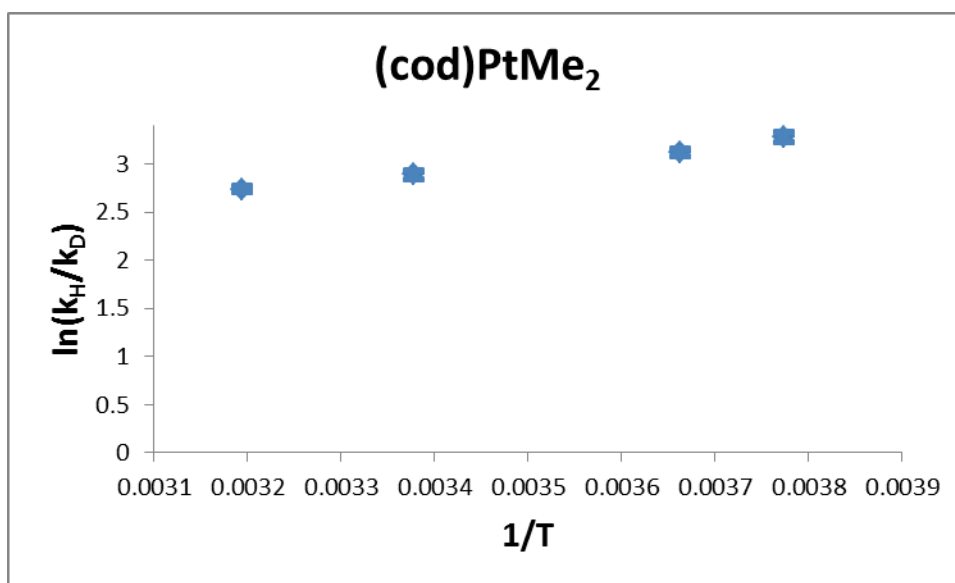
Complex	Temperature	Method	# of runs	KIE
(cod)PtMe ₂	-8	1	7	20.8 ± 1.0
		2	4	26.7 ± 1.4
	0	1	3	20.0 ± 2.1
		2	5	22.7 ± 0.9
		3	3	22.9 ± 1.7
	23	1	4	18.0 ± 0.9
^t Bu ² Cp ₂ ZrMe ₂	40	1	3	16.4 ± 0.6
		2	3	15.5 ± 0.5
	-8	1	3	11.5 ± 1.8
		2	3	10.7 ± 0.2
		2	3	10.4 ± 0.7
	23	1	3	10.0 ± 1.1
		2	2	10.2 ± 0.11
	40	1	8	9.2 ± 0.6
		2	6	9.6 ± 1.0

ZnMe₂	-8	1	3	2.2 ± 0.1
		2	3	3.4 ± 0.3
	0	2	3	3.1 ± 0.3
	23	1	3	3.1 ± 0.7
	40	2	3	2.9 ± 0.3
(dppe)PdMe₂ - TFA	-8	2	4	9.7 ± 0.4
	0	2	4	8.1 ± 0.8
	23	2	4	6.9 ± 0.2
	40	2	4	7.2 ± 0.2
(PONOP)RhMe	-8	1	3	6.0 ± 0.9
	0	2	3	10.5 ± 3.1
	23	1	5	6.1 ± 0.4
(PONOP)IrMe	-8	1	2	13.4 ± 0.8
	23	1	6	10.5 ± 1.1

2. General considerations for Arrhenius plots

All equations and errors of Arrhenius plots were generated using a linear least squares regression with a 95% confidence level by Matlab.

3. Arrhenius plot and equation for (cod)PtMe₂



$$cf(x) = p1 * x + p2$$

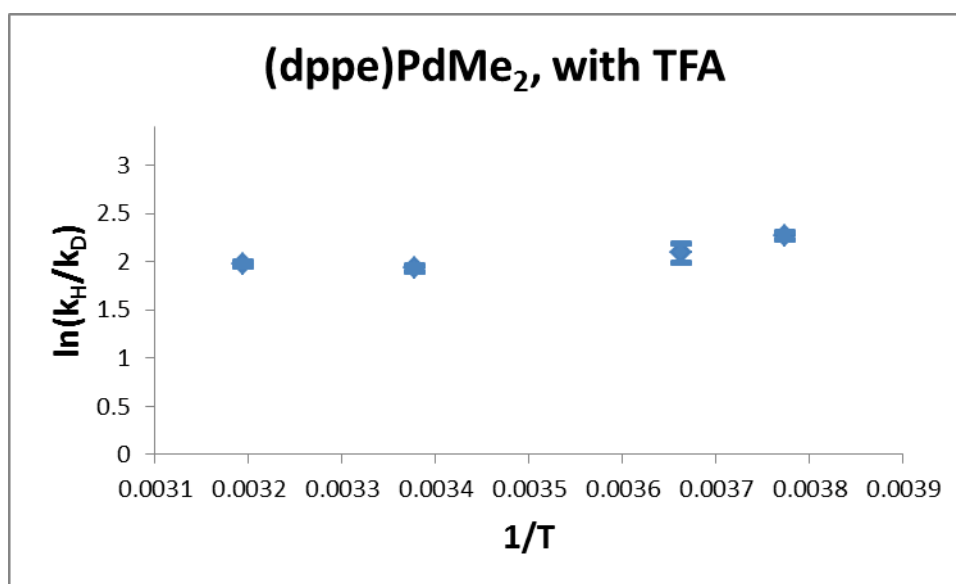
Coefficients (with 95% confidence bounds):

$$p1 = 856.5 \text{ (425.4, 1288)}$$

$$p2 = -0.006875 \text{ (-1.535, 1.521)}$$

goodness of fit (R squared) = 0.97337

4. Arrhenius plot and equation for (dppe)PdMe₂ with TFA



$$cf(x) = p1*x + p2$$

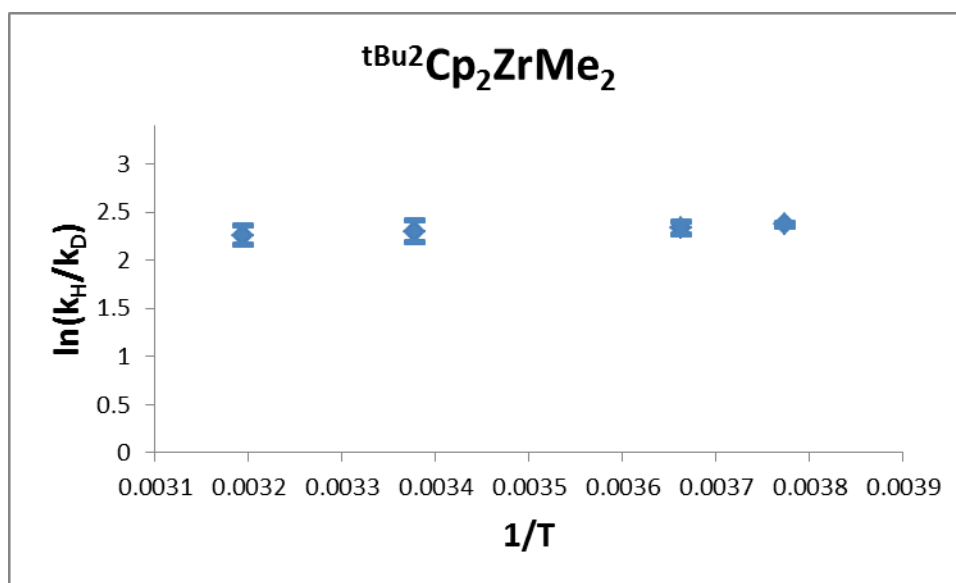
Coefficients (with 95% confidence bounds):

$$p1 = 479.8 \text{ } (-385, 1345)$$

$$p2 = 0.3765 \text{ } (-2.717, 3.47)$$

goodness of fit (R squared) = 0.74023

5. Arrhenius plot and equation for $t\text{Bu}_2\text{Cp}_2\text{ZrMe}_2$



$$cf(x) = p1 * x + p2$$

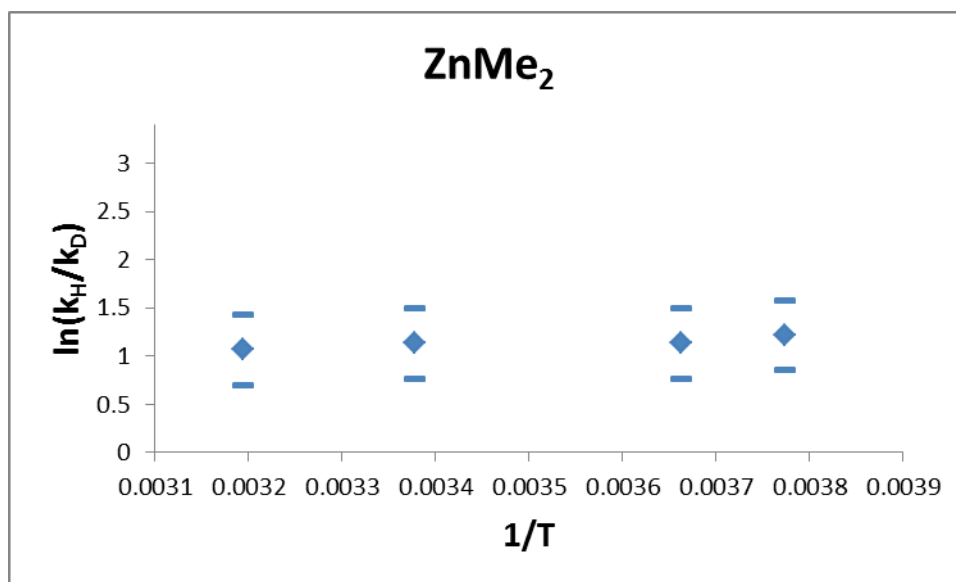
Coefficients (with 95% confidence bounds):

$$p1 = 172.6 \text{ (111.2, 234)}$$

$$p2 = 1.714 \text{ (1.494, 1.934)}$$

goodness of fit (R squared) = 0.98651

6. Arrhenius plot and equation for ZnMe_2



$$cf(x) = p1 * x + p2$$

Coefficients (with 95% confidence bounds):

$$p1 = 178.5 \text{ } (-220.2, 577.1)$$

$$p2 = 0.5181 \text{ } (-0.8668, 1.903)$$

goodness of fit (R squared) = 0.64977

7. Table of parameters and R^2 values for Arrhenius plots for complexes

Complex	$E_a^D - E_a^H$ (kcal)	A_H/A_D	R^2
(cod)PtMe ₂ – fixed error on reported data	3.2 ± 0.2	$0.08 \pm_{-0.02}^{+0.03}$	0.99715
(cod)PtMe ₂	1.7 ± 0.9	$0.9 \pm_{-0.8}^{+3.6}$	0.97337
(dppe)PdMe ₂ - fixed error on reported data using TFE	2.5 ± 0.4	$0.3 \pm_{-0.1}^{+0.3}$	0.99715
(dppe)PdMe ₂ – using TFA	1.0 ± 1.7	$1.5 \pm_{-1.4}^{+34.0}$	0.74023
^t Bu ₂ Cp ₂ ZrMe ₂	0.3 ± 0.1	$5.6 \pm_{-1.1}^{+1.3}$	0.98651
ZnMe ₂	0.4 ± 0.8	$1.7 \pm_{-1.3}^{+5.0}$	0.64977